

Acetonide (**2d**) of Compound **2**. Prepared by treatment of **2** with  $\text{Me}_2\text{CO}$ , *p*-toluenesulphonic acid. Colourless oil. EIMS  $m/z$ : 558  $[\text{M}^+]$ .  $^1\text{H}$  NMR (400 MHz):  $\delta$  0.73, 0.94 (each s, 3H), 0.95 (*d*,  $J = 6.5$  Hz), 1.01, 1.16, 1.23, 1.27 (each s, 3H), 1.39, 1.51 (each s, 3H, Acetonide diMe), 2.61 (s, 1H, H-18), 3.60 (s, 3H,  $-\text{COOMe}$ ), 3.58 (*d*,  $J = 4.1$  Hz, 1H, H-1 or 3), 3.87 (*d*,  $J = 7.7$  Hz, 1H, H-3 or 1), 4.18 (*dd*,  $J = 4.1, 7.7$  Hz, 1H, H-2), 5.40 (*br s*, 1H, H-12). Another acetonide (**2e**) was detected by TLC but not identified with the spectral data.

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TRITERPENOIDS FROM *SALVIA PRZEWALSKII*

NING WANG, MASATAKE NIWA\* and HOU-WEI LUO

Nanjing College of Pharmacy, Nanjing, China; \*Faculty of Pharmacy, Meijo University, Tempaku, Nagoya 468, Japan

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**Key Word Index**—*Salvia przewalskii*; Labiatae; triterpenoids; przewanoic acid A; przewanoic acid B.

**Abstract**—Oleanolic acid and two novel triterpenoids, przewanoic acid A and przewanoic acid B, were isolated from the dried roots of *Salvia przewalskii*. Their structures were elucidated on the basis of chemical and spectral methods.

## INTRODUCTION

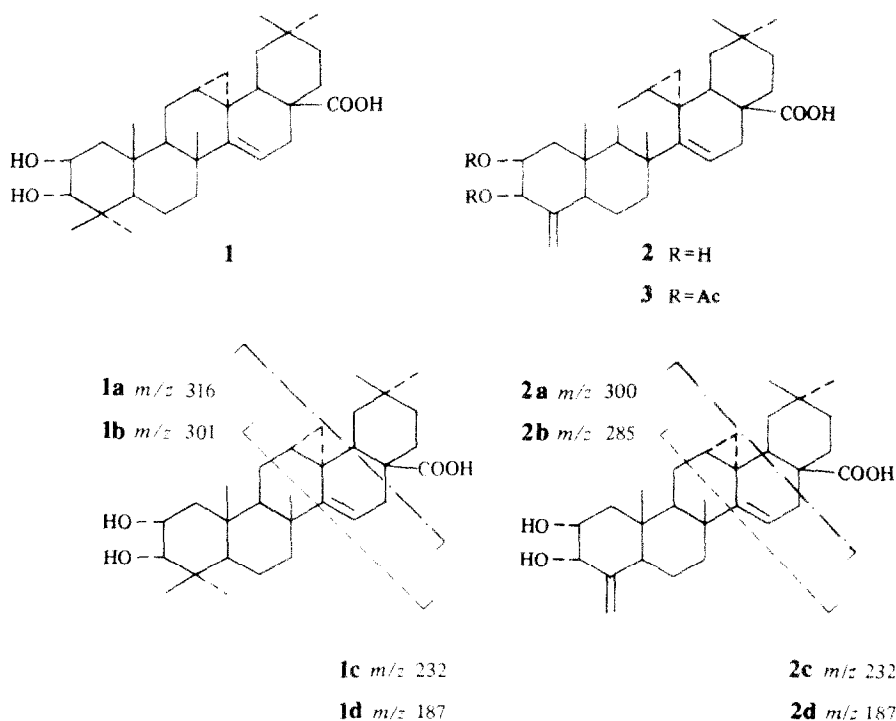
The Tibetan folk drug 'Hong Qin Jiao' is the dried roots of *Salvia przewalskii* Maxim, which is widely distributed in the western areas of China [1–3]. It has been reported that its main chemical components are the *O*-naphthaquinone diterpenes [4, 5]. The present paper describes the isolation and elucidation of two novel triterpenoids, przewanoic acid A and przewanoic acid B, along with a known triterpenoid, oleanolic acid.

## RESULTS AND DISCUSSION

Przewanoic acid A (**1**), white needles, mp 269–270°,  $[\alpha]_D + 125^\circ$  had the molecular formula  $\text{C}_{30}\text{H}_{46}\text{O}_4$ . Its UV spectrum  $\lambda_{\text{max}}^{\text{MeOH}}$  211 nm (log  $\epsilon$  3.65) showed the probable presence of a double bond conjugated cyclopropane [6]. The  $^1\text{H}$  NMR spectrum showed the presence of six methyl groups in agreement with the  $^{13}\text{C}$  NMR spectral data ( $\delta$  17.5, 17.5, 22.2, 22.8, 29.5, 32.6; Table I). Also the  $^1\text{H}$  NMR spectrum revealed one allylic hydrogen and two secondary hydroxyls. The signal at high field ( $-\text{O}H$ , *dd*, 1H,  $J = 4.8, 4.8$  Hz) is characteristic of the  $\text{CH}_2$  in a cyclopropane [6, 7].

The coupling relationships of related protons were assigned from the comparative study of the  $^1\text{H}$ – $^1\text{H}$  2D COSY data. The signals at 5.82 (*dd*, 1H,  $J = 7.2, 3.6$  Hz), 2.79 (*dd*, 1H,  $J = 13.2, 7.2$  Hz) and 2.07 (*dd*, 1H,  $J = 13.2, 3.6$  Hz) were a group of corresponding protons. This showed the presence of part structure  $^{14}\text{C} = ^{15}\text{CH} - ^{16}\text{CH}_2$ . The signals at 4.29 (*ddd*, 1H,  $J = 10.7, 4.0, 2.6$  Hz,  $\text{CHOH}$ ) and 3.77 (*d*, 1H,  $J = 2.6$  Hz,  $\text{CHOH}$ ) were coupled with each other. This indicated two secondary hydroxy groups which should be placed at C-2 and C-3, respectively, in a *trans*-diaxial configuration ( $J_{2\text{H}-3\text{H}} = 2.6$  Hz).

The position of the carboxyl group at C-17 and the double bond at the  $\Delta^{14}$ -position was established from the mass spectral fragmentation pattern of przewanoic acid A (**1**). Compound **1** exhibited a fragment peak at  $m/z$  316 (**1a**). This ion peak was accompanied by a peak 15 mass units lower (**1b**) which was formed by the loss of the allylically activated methyl group at C-8. Moreover, the mass spectrum of **1** showed a peak at  $m/z$  232 (**1c**) derived from rings D and E. Furthermore, the fragment **1c** loses the carboxyl substituent at C-17 giving rise to a fragment base peak at  $m/z$  187 (**1d**). This type of fragmentation is



consistent with the mass spectral data of  $\Delta^{14}$ -taraxerene derivatives reported by Djerassi *et al.* [8–10].

From these data the compound **1** was assigned tentatively the structure 2 $\alpha$ ,3 $\alpha$ -dihydroxy-12,13-cyclo-taraxerane-14-en-28-oic acid.

Another triterpenoid, przewanoic acid B (**2**), white needles, mp 258–259°,  $[\alpha]_D^{25} + 103^\circ$  had the molecular formula  $C_{29}H_{42}O_4$ . Its spectra are similar to those of compound **1**. However, notable differences between **1** and **2** could be seen with regard to the following points. The  $^1H$ NMR spectrum showed the presence of four tertiary methyl groups on saturated carbons at  $\delta$  0.72, 0.90, 0.91, 0.96 (each s and 3 H). Furthermore, the signals at 5.06 and 4.73 (each s and 1 H) showed the presence of an exocyclic methylene group. This was confirmed by the  $^{13}C$ NMR spectral data at  $\delta$  14.5, 22.1, 29.3, 32.3 (each q) and 150.2 (s), 111.3 (d) (Table I).

The position of the exocyclic methylene at C-4 was established by comparison with the HRMS data of compounds **1** and **2**. The fragment at  $m/z$  300.2062 (**2a**,  $C_{20}H_{28}O_2$ ) and 285.1848 (**2b**,  $C_{19}H_{25}O_2$ , **2a**-Me) resulted from RDA cleavage of ring D and are 16 mass units lower than **1a** and **1b**, respectively. On the other hand, both the fragment peaks at  $m/z$  232.1473 (**2c**,  $C_{15}H_{20}O_2$ ) and 187.1462 (**2d**,  $C_{14}H_{19}$ , **2c**-COOH) resulting from ring C cleavage were the same as **1c** and **1d**. Moreover, the chemical shift value of the  $3\beta$ -proton moved from  $\delta$  3.77 in **1** to 4.18 in **2**, due to the deshielding effect of an exocyclic methylene instead of a geminal dimethyl group.

#### EXPERIMENTAL

**Plant material.** The roots of *Salvia przewalskii* Maxim were collected in the Gansu province of China, during the autumn of 1984.

**Extraction.** The dried roots (10 kg) were extracted with hot 95% EtOH. The residue (203 g) after evapn of solvent was chromatographed on a silica gel column eluted successively with cyclohexane  $CH_2Cl_2$  EtOAc, yielding oleanolic acid (25 mg), przewanoic acid A (18 mg) and przewanoic acid B (42 mg) as well as 12 *O*-naphthaquinone diterpenes.

**Przewanoic acid A (1).** Recrystallized from  $Me_2CO$  as white needles, mp 269–270°,  $[\alpha]_D^{25} + 125^\circ$  (MeOH;  $c$  0.08). Found:

Table I.  $^{13}C$ NMR spectra of compounds **1** and **2** (25 MHz, TMS as int. standard)\*

C	1	2	C	1	2
1	42.8 t	42.3 t	16	31.1 t	30.6 t
2	66.2 d	69.2 d	17	52.8 s	52.8 s
3	79.2 d	75.5 d	18	34.9 d	34.5 d
4	38.7 s	150.2 s	19	38.9 t	35.4 t
5	48.7 d	45.1 d	20	29.0 s	28.8 s
6	19.5 t	20.4 t	21	34.3 t	33.8 t
7	35.7 t	36.7 t	22	31.1 t	32.3 t
8	38.7 s	37.9 s	23	29.5 q	111.3 t
9	48.0 d	44.9 d	24	17.5 q	
10	37.5 s	37.2 s	25	17.5 q	14.5 q
11	18.5 t	20.1 t	26	22.2 q	29.3 q
12	14.9 d	14.6 d	27	11.5 t	11.3 t
13	23.8 s	23.4 s	28	179.9 s	180.9 s
14	156.6 s	156.4 s	29	32.6 q	32.3 q
15	118.3 d	117.8 d	30	22.8 q	22.1 q

\*Compound **1** in pyridine, **2** in  $CDCl_3$ .

$[M]^+$  470.3485;  $C_{30}H_{46}O_4$  requires: 470.3564. UV  $\lambda_{\text{Max}}^{\text{MeOH}}$  nm (log  $\epsilon$ ): 211 (3.65). IR  $\nu_{\text{Max}}^{\text{KBr}}$   $\text{cm}^{-1}$ : 3584, 3384, 1695.  $^1\text{H}$  NMR (400 MHz, pyridine- $d_5$ ):  $\delta$  5.82 (dd, 1H,  $J = 7.2, 3.6$  Hz), 4.29 (ddd, 1H,  $J = 10.7, 4.0, 2.6$  Hz), 3.77 (d, 1H,  $J = 2.6$  Hz), 0.86, 0.93, 0.94, 1.04, 1.13, 1.20 (each 3H, s), -0.08 (dd, 1H,  $J = 4.8, 4.8$  Hz). For  $^{13}\text{C}$  NMR, see Table 1. EIMS (70 eV)  $m/z$  (rel. int.): 470  $[M]^+$  (82.8), 455  $[M - \text{Me}]^+$  (7.5), 452  $[M - \text{H}_2\text{O}]^+$  (90.9), 425  $[M - \text{COOH}]^+$  (7.7), 316 (2.1), 301 (30.5), 232 (1.2), 187 (100), 133 (34.2).

**Przewanoic acid B (2).** Recrystallized from MeOH as white needles, mp 258–259°,  $[\alpha]_D + 103^\circ$  (MeOH;  $c$  0.465). Found:  $[M]^+$  454.3381;  $C_{29}H_{42}O_4$  requires: 454.3252. UV  $\lambda_{\text{Max}}^{\text{MeOH}}$  nm (log  $\epsilon$ ): 210 (3.65). IR  $\nu_{\text{Max}}^{\text{KBr}}$   $\text{cm}^{-1}$ : 3426, 1694, 1650, 905.  $^1\text{H}$  NMR (400 MHz  $\text{CDCl}_3$ ):  $\delta$  5.58 (dd, 1H,  $J = 7.4, 4.0$  Hz), 5.06 (s, 1H), 4.73 (s, 1H), -0.08 (dd, 1H,  $J = 4.8$  Hz). For  $^{13}\text{C}$  NMR, see Table 1. EIMS (70 eV)  $m/z$  (rel. int.): 454  $[M]^+$  (29.1), 439  $[M - \text{Me}]^+$  (5.1), 436  $[M - \text{H}_2\text{O}]^+$  (34.8), 409  $[M - \text{COOH}]^+$  (4.5), 300 (8.2), 285 (8.6), 232 (4.1), 187 (100), 133 (22.1).

**Przewanoic acid B acetate (3).** Compound **2** (6 mg) was dissolved in pyridine (0.4 ml) and  $\text{Ac}_2\text{O}$  (0.6 ml) and left overnight. Normal work-up gave przewanoic acid B acetate (4 mg) as needles.  $^1\text{H}$  NMR (90 MHz  $\text{CDCl}_3$ ):  $\delta$  1.98 (s, 3H, COMe), 2.07 (s, 3H, COMe).

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## MYRIANTHINIC ACID: A NEW TRITERPENOID FROM MYRIANTHUS ARBOREUS

F. NGNINZEKO NGOUNOU, DAVID LONTSI and B. LUCAS SONDENGAM\*

Department of Organic Chemistry, University of Yaoundé, D.P. 812 Yaoundé, Cameroon.

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**Key Word Index**—*Myrianthus arboreus*; Cecropiaceae; bark; pentacyclic triterpenoids; Myrianthnic acid;  $3\beta,6\beta$ -dihydroxyolean-12-en-29-oic acid.

**Abstract**—A new pentacyclic triterpene acid has been isolated from the stem bark of *Myrianthus arboreus* and its structure has been established as  $3\beta,6\beta$ -dihydroxyolean-12-en-29-oic acid and named myrianthnic acid.

## INTRODUCTION

*Myrianthus arboreus* P. Beauv., Cecropiaceae [1] is a small tropical African tree, which grows from Guinea to southern Angola, from southern Sudan to western Tanzania and through Ouganda. The plant is widely used in indigenous medicine: the infusion of bark to treat dysentery; the leaves against heartaches, the accidents during pregnancy and dysmenorrhoea; the juice of young leaves against toothaches and bronchitis [2]. Early works on *M. arboreus* reported the isolation of peptide alkaloids from the leaves [3], tormentic acid, 2-acetyl tormentic

acid, 3-acetyl tormentic acid and euscaphic acid from the root wood [4, 5], and Myrianthnic acid from the same organ [6]. We now report the isolation from the trunk bark of *M. arboreus*, of a new triterpenic acid, myrianthnic acid (2) as its methyl ester (1).

## RESULTS

From the methylated ethyl acetate extracts of ground barks, the methyl ester of myrianthnic acid (1)  $\text{C}_{31}\text{H}_{50}\text{O}_4$  was obtained which crystallized from methylene chloride as colourless granules, mp 145–147°. The IR spectrum revealed absorptions at  $\nu_{\text{max}}$  1700 ( $-\text{COOMe}$ ) and 1645  $\text{cm}^{-1}$ , and at 3450 and 3525  $\text{cm}^{-1}$ , indicative of the presence of two hydroxyl groups. Myrianthnic acid

\*Author to whom correspondence should be addressed.